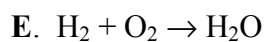
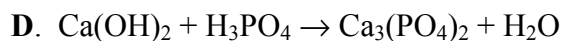
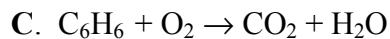
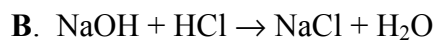
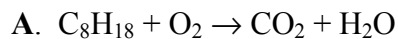


### *Stoichiometry III*

Name: \_\_\_\_\_

Score: \_\_\_\_\_

Consider the five **unbalanced** chemical reactions below. Imagine 10.00g of both reactants is utilized. Rank these species from greatest to least number of moles of water produced in the reaction.



Greatest 1 \_\_\_\_\_ 2 \_\_\_\_\_ 3 \_\_\_\_\_ 4 \_\_\_\_\_ 5 \_\_\_\_\_ Least

Explain your reasoning and show all work below.

Circle the response that best describes your confidence in your answer above.

(Basically Guessed) 1      2      3      4      5 (Positive you get it)